### Amerex Fire Pacific Pty Ltd

Chemwatch: 4887-11 Version No: 2.1.1.1

Safety Data Sheet according to WHS and ADG requirements

Chemwatch Hazard Alert Code: 2

Issue Date: 03/31/2014 Print Date: 06/21/2016 Initial Date: Not Available LIGHS AUS EN

### SECTION 1 IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

Product Identifier			
Product name	ABE Powder Fire Extinguisher		
Synonyms	Not Available		
Proper shipping name	FIRE EXTINGUISHERS with compressed or liquefied gas		
Other means of identification	Not Available		
Relevant identified us	es of the substance or mixture and uses advised against		
Relevant identified uses	Fire extinguishing agent.		
Registered company	Amerex Fire Pacific Pty Ltd		
name Address	2A Staple Street, Seventeen Mile Rocks QLD 4073 Australia		
Telephone	+61 7 3715 5644		
Fax	+61 7 3715 8450		
Website	Not Available		
Email	Not Available		
Emergency telephone	e number		
Association / Organisation	Not Available		
Emergency telephone numbers	+61 7 3715 5644 (Mon-Fri 9am-5pm)		
Other emergency telephone numbers	Not Available		

### **SECTION 2 HAZARDS IDENTIFICATION**

### Classification of the substance or mixture

HAZARDOUS CHEMICAL. DANGEROUS GOODS. According to the WHS Regulations and the ADG Code.

### CHEMWATCH HAZARD RATINGS

Min Flammability 0 Toxicity 1 0 = Minimum 1 = Low 2 = Moderate 3 = High 4 = Extreme **Body Contact** 2 Reactivity 1 Chronic 0

Poisons Schedule	Not Applicable
Classification <sup>[1]</sup>	Skin Corrosion/Irritation Category 2, Eye Irritation Category 2A, Specific target organ toxicity - single exposure Category 3 (respiratory tract irritation), Chronic Aquatic Hazard Category 4
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HSIS; 3. Classification drawn from EC Directive 1272/2008 - Annex

### Label elements

### **GHS label elements**



SIGNAL WORD

WARNING

### Hazard statement(s)

H315	Causes skin irritation.	
H319	Causes serious eye irritation.	
H335	May cause respiratory irritation.	
H413	May cause long lasting harmful effects to aquatic life.	
AUH044	Risk of explosion if heated under confinement	

### Precautionary statement(s) Prevention

P271	Use only outdoors or in a well-ventilated area.
P261	Avoid breathing dust/fumes.
P273	Avoid release to the environment.
P280	Wear protective gloves/protective clothing/eye protection/face protection.

### Precautionary statement(s) Response

Take off contaminated clothing and wash before reuse.	
IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue sing.	
all a POISON CENTER or doctor/physician if you feel unwell.	
f eye irritation persists: Get medical advice/attention.	
ON SKIN: Wash with plenty of soap and water.	
IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing.	
If skin irritation occurs: Get medical advice/attention.	

### Precautionary statement(s) Storage

	(-)
P405	Store locked up.
P403+P233	Store in a well-ventilated place. Keep container tightly closed.

# Precautionary statement(s) Disposal P501 Dispose of content

1	Dispose of contents/container in accordance with local regulations.

### SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

### Substances

See section below for composition of Mixtures

### **Mixtures**

CAS No	%[weight]	Name
7722-76-1	>60	ammonium phosphate, monobasic
Not Available	NotSpec.	propellant as,
7727-37-9.	NotSpec.	nitrogen

### **SECTION 4 FIRST AID MEASURES**

### Description of first aid measures

Eye	Contact
-----	---------

Generally not applicable.

If skin contact occurs:

Skin Contact | Immediately remove all contaminated clothing, including footwear.

Issue Date: 03/31/2014 Print Date: 06/21/2016

- Flush skin and hair with running water (and soap if available).
- Seek medical attention in event of irritation.
- Following exposure to gas, remove the patient from the gas source or contaminated area,
- NOTE: Personal Protective Equipment (PPE), including positive pressure self-contained breathing apparatus may be required to assure the safety of the rescuer.
- Prostheses such as false teeth, which may block the airway, should be removed, where possible, prior to initiating first aid procedures.
- If the patient is not breathing spontaneously, administer rescue breathing.
- If the patient does not have a pulse, administer CPR.
- ▶ If medical oxygen and appropriately trained personnel are available, administer 100% oxygen.
- Summon an emergency ambulance. If an ambulance is not available, contact a physician, hospital, or Poison Control Centre for further instruction.
- Keep the patient warm, comfortable and at rest while awaiting medical care.
- MONITOR THE BREATHING AND PULSE, CONTINUOUSLY.
- Administer rescue breathing (preferably with a demand-valve resuscitator, bag-valve mask-device, or pocket mask as trained) or CPR if necessary.

Ingestion

Inhalation

▶ Not considered a normal route of entry.

### Indication of any immediate medical attention and special treatment needed

For gas exposures:
-----BASIC TREATMENT

• Establish a patent airway with suction where necessary.

- · Watch for signs of respiratory insufficiency and assist ventilation as necessary.
- ▶ Administer oxygen by non-rebreather mask at 10 to 15 l/min.
- · Monitor and treat, where necessary, for pulmonary oedema .
- · Monitor and treat, where necessary, for shock.
- Anticipate seizures.

ADVANCED TREATMENT

- · Consider orotracheal or nasotracheal intubation for airway control in unconscious patient or where respiratory arrest has occurred.
- ▶ Positive-pressure ventilation using a bag-valve mask might be of use.
- Monitor and treat, where necessary, for arrhythmias.
- Start an IV D5W TKO. If signs of hypovolaemia are present use lactated Ringers solution. Fluid overload might create complications.
- Drug therapy should be considered for pulmonary oedema.
- ▶ Hypotension with signs of hypovolaemia requires the cautious administration of fluids. Fluid overload might create complications.
- Treat seizures with diazepam.
- ▶ Proparacaine hydrochloride should be used to assist eye irrigation.

BRONSTEIN, A.C. and CURRANCE, P.L.

EMERGENCY CARE FOR HAZARDOUS MATERIALS EXPOSURE: 2nd Ed. 1994

### **SECTION 5 FIREFIGHTING MEASURES**

### **Extinguishing media**

Non flammable.

### Special hazards arising from the substrate or mixture

Fire Incompatibility

 Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result

### Advice for firefighters

Fire/Explosion Hazard

Fire Fighting

If safe to do so, remove containers from path of fire.

Cool fire exposed containers with water spray from a protected location.

- Containers may explode when heated Ruptured cylinders may rocket
- Fire exposed containers may vent contents through pressure relief devices.
- High concentrations of gas may cause asphyxiation without warning.
- May decompose explosively when heated or involved in fire.
- · Contact with gas may cause burns, severe injury and/ or frostbite.

Decomposition may produce toxic fumes of; carbon monoxide (CO) carbon dioxide (CO2) nitrogen oxides (NOx) phosphorus oxides (POx) other pyrolysis products typical of burning organic material

### **SECTION 6 ACCIDENTAL RELEASE MEASURES**

### Personal precautions, protective equipment and emergency procedures

Minor Spills

• Avoid breathing vapour and any contact with liquid or gas. Protective equipment including respirator should be used.

Issue Date: 03/31/2014 Print Date: 06/21/2016

- ▶ DO NOT enter confined spaces were gas may have accumulated.
- ▶ Increase ventilation.
- · Clear area of personnel.
- ▶ Stop leak only if safe to so do.
- ▶ Remove leaking cylinders to safe place. Release pressure under safe controlled conditions by opening valve.
- ▶ Do not exert excessive pressure on the valve; do not attempt to operate a damaged valve
- Orientate cylinder so that the leak is gas, not liquid, to minimise rate of leakage
- · Keep area clear of personnel until gas has dispersed.
- · Clean up all spills immediately.
- Wear protective clothing, safety glasses, dust mask, gloves.
- ▶ Secure load if safe to do so. Bundle/collect recoverable product.
- Use dry clean up procedures and avoid generating dust.
- Vacuum up (consider explosion-proof machines designed to be grounded during storage and use).
- Water may be used to prevent dusting.
- Collect remaining material in containers with covers for disposal.
- Flush spill area with water.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

#### SECTION 7 HANDLING AND STORAGE

### Precautions for safe handling

**Major Spills** 

- Consider use in closed pressurised systems, fitted with temperature, pressure and safety relief valves which are vented for safe dispersal.
- ► The tubing network design connecting gas cylinders to the delivery system should include appropriate pressure indicators and vacuum or suction lines.
- Fully-welded types of pressure gauges, where the bourdon tube sensing element is welded to the gauge body, are recommended.
- Before connecting gas cylinders, ensure manifold is mechanically secure and does not containing another gas. Before
  disconnecting gas cylinder, isolate supply line segment proximal to cylinder, remove trapped gas in supply line with aid of
  vacuum pump
- ▶ When connecting or replacing cylinders take care to avoid airborne particulates violently ejected when system pressurises.
- Consider the use of doubly-contained piping; diaphragm or bellows sealed, soft seat valves; backflow prevention devices; flash arrestors; and flow monitoring or limiting devices. Gas cabinets, with appropriate exhaust treatment, are recommended, as is automatic monitoring of the secondary enclosures and work areas for release.
- ▶ Use a pressure reducing regulator when connecting cylinder to lower pressure (<100 psig) piping or systems
- ▶ Use a check valve or trap in the discharge line to prevent hazardous back-flow into the cylinder
- Check regularly for spills or leaks. Keep valves tightly closed but do not apply extra leverage to hand wheels or cylinder kevs.
- Open valve slowly. If valve is resistant to opening then contact your supervisor
- Valve protection caps must remain in place must remain in place unless container is secured with valve outlet piped to use point.
- Never insert a pointed object (e.g hooks) into cylinder cap openings as a means to open cap or move cylinder. Such action can inadvertently turn the valve and gas a gas leak. Use an adjustable strap instead of wrench to free an over-tight or rusted cap.
- A bubble of gas may buildup behind the outlet dust cap during transportation, after prolonged storage, due to defective cylinder valve or if a dust cap is inserted without adequate evacuation of gas from the line. When loosening dust cap, preferably stand cylinder in a suitable enclosure and take cap off slowly. Never face the dust cap directly when removing it; point cap away from any personnel or any object that may pose a hazard, under negative pressure (relative to atmospheric gas)
- Do NOT drag, slide or roll cylinders use a suitable hand truck for cylinder movement
- ▶ Test for leakage with brush and detergent NEVER use a naked flame.
- ▶ Do NOT heat cylinder by any means to increase the discharge rate of product from cylinder.
- Leaking gland nuts may be tightened if necessary.
- If a cylinder valve will not close completely, remove the cylinder to a well ventilated location (e.g. outside) and, when empty, tag as FAULTY and return to supplier.
- Obtain a work permit before attempting any repairs.
- DO NOT attempt repair work on lines, vessels under pressure.
- Atmospheres must be tested and O.K. before work resumes after leakage.

### DO NOT transfer gas from one cylinder to another.

- ► Cylinders should be stored in a purpose-built compound with good ventilation, preferably in the open.
- ▶ Such compounds should be sited and built in accordance with statutory requirements.
- ▶ The storage compound should be kept clear and access restricted to authorised personnel only.
- · Cylinders stored in the open should be protected against rust and extremes of weather.
- Cylinders stored in the open should be protected against rust and extremes of wea
   Cylinders in storage should be properly secured to prevent toppling or rolling.
- Cylinders in storage should be properly secured to prevent
   Cylinder valves should be closed when not in use.
- Where cylinders are fitted with valve protection this should be in place and properly secured.

# Safe handling

## Other information

### Issue Date: 03/31/2014 Print Date: 06/21/2016

### ABE Powder Fire Extinguisher

- Gas cylinders should be segregated according to the requirements of the Dangerous Goods Act.
- · Preferably store full and empty cylinders separately.
- · Check storage areas for hazardous concentrations of gases prior to entry.
- Full cylinders should be arranged so that the oldest stock is used first,
- Cylinders in storage should be checked periodically for general condition and leakage.
- Protect cylinders against physical damage. Move and store cylinders correctly as instructed for their manual handling.

NOTE: A 'G' size cylinder is usually too heavy for an inexperienced operator to raise or lower.

### Conditions for safe storage, including any incompatibilities

- · Glass container is suitable for laboratory quantities
- Cylinder:
- Ensure the use of equipment rated for cylinder pressure.
- Ensure the use of compatible materials of construction.
- Valve protection cap to be in place until cylinder is secured, connected.
  - · Cylinder must be properly secured either in use or in storage.
  - Cylinder valve must be closed when not in use or when empty.
  - ▶ Segregate full from empty cylinders.

WARNING: Suckback into cylinder may result in rupture. Use back-flow preventive device in piping.

### Storage incompatibility

Suitable container

- · Avoid strong bases.
- Compressed gases may contain a large amount of kinetic energy over and above that potentially available from the energy of reaction produced by the gas in chemical reaction with other substances

### SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

### Control parameters

### OCCUPATIONAL EXPOSURE LIMITS (OEL)

#### INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
Australia Exposure Standards	nitrogen	Nitrogen	Not Available	Not Available	Not Available	Asphyxiant

### **EMERGENCY LIMITS**

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
ammonium phosphate, monobasic	Ammonium dihydrogen phosphate; (Monoammonium phosphate)	12 mg/m3	130 mg/m3	790 mg/m3
nitrogen	Nitrogen	7.96E+05 ppm	832000 ppm	869000 ppm

Ingredient	Original IDLH	Revised IDLH
ammonium phosphate, monobasic	Not Available	Not Available
propellant as,	Not Available	Not Available
nitrogen	Not Available	Not Available

### MATERIAL DATA

### **Exposure controls**

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match the particular process and chemical or contaminant in use

## Appropriate engineering controls

Employers may need to use multiple types of controls to prevent employee overexposure.

- Areas where cylinders are stored require good ventilation and, if enclosed, need discrete/controlled exhaust ventilation.
- · Secondary containment and exhaust gas treatment may be required by certain jurisdictions.
- Local exhaust ventilation may be required in work areas.
- Consideration should be given to the use of diaphragm or bellows-sealed, soft-seat valves; backflow prevention devices and flow-monitoring or limiting devices.
- Automated alerting systems with automatic shutdown of gas-flow may be appropriate and may in fact be mandatory in certain jurisdictions,

Issue Date: 03/31/2014 Print Date: 06/21/2016

- Respiratory protection in the form of air-supplied or self-contained breathing equipment must be worn if the oxygen concentration in the workplace air is less than 19%.
- Cartridge respirators do NOT give protection and may result in rapid suffocation.

Air contaminants generated in the workplace possess varying "escape" velocities which, in turn, determine the "capture velocities" of fresh circulating air required to effectively remove the contaminant.

Type of Contaminant:	Air Speed:
gas discharge (active generation into zone of rapid air motion)	1-2.5 m/s (200-500 f/min.)

Within each range the appropriate value depends on:

Lower end of the range	Upper end of the range
1: Room air currents minimal or favourable to capture	1: Disturbing room air currents
2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of high toxicity
3: Intermittent, low production.	3: High production, heavy use
4: Large hood or large air mass in motion	4: Small hood-local control only

Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2.5 m/s (200-500 f/min.) for extraction of gases discharged 2 meters distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.

### Personal protection









- · Safety glasses with side shields.
- · Chemical goggles.

## Eye and face protection

Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]

### Skin protection

See Hand protection below

### Hands/feet protection

When handling sealed and suitably insulated cylinders wear cloth or leather gloves.

Wear chemical protective gloves, e.g. PVC. Wear safety footwear.

### Body protection

See Other protection below

- · Protective overalls, closely fitted at neck and wrist.
- ▶ Eye-wash unit.

### Other protection

- Ensure availability of lifeline in confined spaces.
- Staff should be trained in all aspects of rescue work.
- ▶ Rescue gear: Two sets of SCBA breathing apparatus Rescue Harness, lines etc.

### Thermal hazards

Not Available

### Respiratory protection

- Respirators may be necessary when engineering and administrative controls do not adequately prevent exposures.
- The decision to use respiratory protection should be based on professional judgment that takes into account toxicity information, exposure measurement data, and frequency and likelihood of the worker's exposure ensure users are not subject to high thermal loads which may result in heat stress or distress due to personal protective equipment (powered, positive flow, full face apparatus may be an option).
- Published occupational exposure limits, where they exist, will assist in determining the adequacy of the selected respiratory protection. These may be government mandated or vendor recommended.
- Certified respirators will be useful for protecting workers from inhalation of particulates when properly selected and fit tested as part of a complete respiratory protection program.
- ▶ Use approved positive flow mask if significant quantities of dust becomes airborne.
- ▶ Try to avoid creating dust conditions.
- Positive pressure, full face, air-supplied breathing apparatus should be used for work in enclosed spaces if a leak is suspected or the primary containment is to be opened (e.g. for a cylinder change)
- · Air-supplied breathing apparatus is required where release of gas from primary containment is either suspected or demonstrated.

Print Date: 06/21/2016

### **SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES**

Appearance	Pale blue powder with a weak of	dour; slightly soluble in water.	
Physical state	Manufactured	Relative density (Water = 1)	~0.85
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Applicable
pH (as supplied)	Not Applicable	Decomposition temperature	Not Available
Melting point / freezing point (°C)	>300	Viscosity (cSt)	Not Applicable
Initial boiling point and boiling range (°C)	Not Applicable	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	Not Applicable	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Applicable	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Applicable	Surface Tension (dyn/cm or mN/m)	Not Applicable
Lower Explosive Limit (%)	Not Applicable	Volatile Component (%vol)	Not Applicable
Vapour pressure (kPa)	Negligible	Gas group	Not Available
Solubility in water (g/L)	Partly Miscible	pH as a solution (1%)	4.6-4.7
Vapour density (Air = 1)	Not Applicable	VOC g/L	Not Available
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### **SECTION 10 STABILITY AND REACTIVITY**

Reactivity	See section 7
Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable.</li> <li>Hazardous polymerisation will not occur.</li> </ul>
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

### **SECTION 11 TOXICOLOGICAL INFORMATION**

### Information on toxicological effects

Evidence shows, or practical experience predicts, that the material produces irritation of the respiratory system, in a substantial number of individuals, following inhalation. In contrast to most organs, the lung is able to respond to a chemical insult by first removing or neutralising the irritant and then repairing the damage. The repair process, which initially evolved to protect mammalian lungs from foreign matter and antigens, may however, produce further lung damage resulting in the impairment of gas exchange, the primary function of the lungs. Respiratory tract irritation often results in an inflammatory response involving the recruitment and activation of many cell types, mainly derived from the vascular system. Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may be damaging to the health of the individual.

Inhaled

- Common, generalised symptoms associated with non-toxic gas inhalation include : • central nervous system effects such as headache, confusion, dizziness, progressive stupor, coma and seizures;
- · respiratory system complications may include tachypnoea and dyspnoea;
- cardiovascular effects may include circulatory collapse and arrhythmias;
- pastrointestinal effects may also be present and may include mucous membrane irritation and nausea and vomiting.

sensitisation

Mutagenicity

### ABE Powder Fire Extinguisher

Ingestion Considered an unlikely route of entry in commercial/industrial environments Evidence exists, or practical experience predicts, that the material either produces inflammation of the skin in a substantial number of individuals following direct contact, and/or produces significant inflammation when applied to the healthy intact skin of animals, for up to four hours, such inflammation being present twenty-four hours or more after the end of the exposure period. Skin irritation may also be present after prolonged or repeated exposure; this may result in a form of contact dermatitis (nonallergic). The dermatitis is often characterised by skin redness (erythema) and swelling (oedema) Skin Contact which may progress to blistering (vesiculation), scaling and thickening of the epidermis. At the microscopic level there may be intercellular oedema of the spongy layer of the skin (spongiosis) and intracellular oedema of the epidermis. The material may accentuate any pre-existing dermatitis condition Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream through, for example, cuts, abrasions, puncture wounds or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected. Evidence exists, or practical experience predicts, that the material may cause eye irritation in a substantial number of individuals and/or may produce significant ocular lesions which are present twenty-four hours or more after instillation into the eve(s) of experimental animals. Repeated or prolonged eye contact may cause inflammation characterised by temporary redness (similar to windburn) of the Eye conjunctiva (conjunctivitis); temporary impairment of vision and/or other transient eye damage/ulceration may occur. Direct contact with the eye may not cause irritation because of the extreme volatility of the gas; however concentrated atmospheres may produce irritation after brief exposures.. Limited evidence suggests that repeated or long-term occupational exposure may produce cumulative health effects involving organs or biochemical systems Principal route of occupational exposure to the gas is by inhalation. Chronic Dogs given daily doses of sodium phosphate dibasic for 9-22 weeks showed calcium deposits in the kidneys (nephrocalcinosis) with disseminated atrophy of the proximal tubule. Animals fed on sodium phosphate dibasic and potassium dihydrogen phosphate, in both short- and long-term studies, showed increased bone porosity; hyperparathyroidism and soft tissue calcification were also evident. TOXICITY IRRITATION ABE Powder Fire Extinguisher Not Available Not Available TOXICITY IRRITATION ammonium dermal (rat) LD50: >5000 mg/kg[1] Not Available phosphate. monobasic Oral (rat) LD50: >1000 mg/kg[1] TOXICITY IRRITATION nitrogen Not Available Not Available 1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.\* Value obtained from manufacturer's SDS Legend: Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances Asthma-like symptoms may continue for months or even years after exposure to the material ceases. This may be due to a non-allergenic condition known as reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes to hours of a **AMMONIUM** documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe PHOSPHATE, bronchial hyperreactivity on methacholine challenge testing and the lack of minimal lymphocytic inflammation, without MONOBASIC eosinophilia, have also been included in the criteria for diagnosis of RADS. RADS (or asthma) following an irritating inhalation is an infrequent disorder with rates related to the concentration of and duration of exposure to the irritating substance. Industrial bronchitis, on the other hand, is a disorder that occurs as result of exposure due to high concentrations of irritating substance (often particulate in nature) and is completely reversible after exposure ceases. The disorder is characterised by dyspnea, cough and mucus production. No significant acute toxicological data identified in literature search. **NITROGEN** No significant acute toxicological data identified in literature search. **Acute Toxicity** Carcinogenicity Skin Reproductivity Irritation/Corrosion Serious Eve STOT - Single Damage/Irritation Exposure Respiratory or Skin STOT - Repeated

Exposure

**Aspiration Hazard** 

### Issue Date: 03/31/2014 Print Date: 06/21/2016

### ABE Powder Fire Extinguisher

**Legena**:

👗 – Data avaliable but goes not till the criteria for classification

Data required to make classification available

○ – Data Not Available to make classification

### **SECTION 12 ECOLOGICAL INFORMATION**

### **Toxicity**

Ingredient	Endpoint	Test Duration (hr)	Species	Value	Source
ammonium phosphate, monobasic	LC50	96	Fish	>85.9mg/L	2
ammonium phosphate, monobasic	EC50	72	Algae or other aquatic plants	>97.1mg/L	2
ammonium phosphate, monobasic	NOEC	72	Algae or other aquatic plants	3.57mg/L	2
Legend:	3. EPIWIN Suite	V3.12 - Aquatic Toxicity Data	pe ECHA Registered Substances - Ec (Estimated) 4. US EPA, Ecotox databa an) - Bioconcentration Data 7. METI (J	ase - Aquatic Toxicity	Data 5. ECETOC

May cause long-term adverse effects in the aquatic environment.

DO NOT discharge into sewer or waterways.

### Persistence and degradability

Ingredient Persistence: Water/Soil		Persistence: Air
ammonium phosphate, monobasic	HIGH	HIGH

### Bioaccumulative potential

57	
Ingredient	Bioaccumulation
ammonium phosphate, monobasic	LOW (LogKOW = -0.7699)

### Mobility in soil

Ingredient	Mobility
ammonium phosphate, monobasic	HIGH (KOC = 1)

### **SECTION 13 DISPOSAL CONSIDERATIONS**

### Waste treatment methods

Product / Packaging disposal

- Evaporate residue at an approved site.
- Return empty containers to supplier. If containers are marked non-returnable establish means of disposal with manufacturer prior to purchase.
- Ensure damaged or non-returnable cylinders are gas-free before disposal.

### **SECTION 14 TRANSPORT INFORMATION**

### Labels Required



Marine Pollutant

NC

HAZCHEM

Not Applicable

Land transport (ADG)

Print Date: 06/21/2016

UN number	1044		
Packing group	Not Applicable		
UN proper shipping name	FIRE EXTINGUISHERS with compressed or liquefied gas		
Environmental hazard	Not Applicable		
Transport hazard	Class 2.2		
class(es)	Subrisk Not Applicable		
Special precautions	Special provisions 225		
for user	Limited quantity 120 ml		

### Air transport (ICAO-IATA / DGR)

UN number	1044			
Packing group	Not Applicable			
UN proper shipping name	Fire extinguishers with compressed or liquefied gas			
Environmental hazard	Not Applicable			
	ICAO/IATA Class	2.2		
Transport hazard class(es)	ICAO / IATA Subrisk	Not Applicable		
01435(65)	ERG Code	2L		
	Special provisions		A19	
	Cargo Only Packing Instructions		213	
NO.001 SE SE SON	Cargo Only Maximum Qty / Pack		150 kg	
Special precautions for user	Passenger and Cargo Packing Instructions		213	
101 4301	Passenger and Cargo Maximum Qty / Pack		75 kg	
	Passenger and Cargo	Limited Quantity Packing Instructions	Forbidden	
	Passenger and Cargo	Limited Maximum Qty / Pack	Forbidden	

### Sea transport (IMDG-Code / GGVSee)

UN number	1044	
Packing group	Not Applicable	
UN proper shipping name	FIRE EXTINGUIS	HERS with compressed or liquefied gas
Environmental hazard	Not Applicable	
Transport hazard	IMDG Class	2.2
class(es)	IMDG Subrisk	Not Applicable
Special precautions for user	EMS Number	F-C, S-V
	Special provisio	ns 225
	Limited Quantiti	es 120 mL

### Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

### **SECTION 15 REGULATORY INFORMATION**

### Safety, health and environmental regulations / legislation specific for the substance or mixture

AMMONIUM PHOSPHATE, MONOBASIC(7722-76-1) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Australia Inventory of Chemical Substances (AICS)

### NITROGEN(7727-37-9.) IS FOUND ON THE FOLLOWING REGULATORY LISTS

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### ABE Powder Fire Extinguisher

National Inventory	Status
Australia - AICS	Υ
Canada - DSL	Υ
Canada - NDSL	N (nitrogen; ammonium phosphate, monobasic)
China - IECSC	Υ
Europe - EINEC / ELINCS / NLP	Y
Japan - ENCS	N (nitrogen)
Korea - KECI	Υ
New Zealand - NZIoC	Υ
Philippines - PICCS	N (nitrogen)
USA - TSCA	Υ
Legend:	Y = All ingredients are on the inventory  N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

### **SECTION 16 OTHER INFORMATION**

### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

A list of reference resources used to assist the committee may be found at:

www.chemwatch.net

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

### **Definitions and abbreviations**

PC—TWA: Permissible Concentration-Time Weighted Average PC—STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit。

IDLH: Immediately Dangerous to Life or Health Concentrations

OSF: Odour Safety Factor

NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index

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